

## CHEM 180: Chemical Reactivity I

The purpose of this syllabus is to describe the course, resources, and policies. It is meant to help all students understand the expectations and requirements for the course, and it should be used as a reference for questions about policies. When updates to the syllabus are made during the term, a new version will be posted electronically, and all students will be notified.

### Course Information

**Course:** Chemistry 180 – Chemical Reactivity I (3 credits: Lecture & Discussion)

**Prerequisites:** Completion of Math 118 and Chem 160, or the equivalent, with a grade of C- or better. A student missing a prerequisite may be withdrawn at any time.

**Time Zone:** This syllabus lists dates/times using Chicago local time (U.S. Central Time Zone)

**Lectures:** Crown Center Auditorium, Tuesdays, Thursdays, 1:00-2:15pm

**Discussions:** You must attend the section for which you registered:

- Section 002, Mondays, 8:15am, Flanner Hall 105
- Section 003, Mondays, 9:20am, Flanner Hall 105

### Instructor Contact Information

**Section Instructor:** Dr. Sandra Helquist (Ph.D.)

**Office:** Flanner Hall 200-B (shared office suite)

**Email:** shelquist@luc.edu

**Email Policy & Timing:** Use your [Loyola \(@luc.edu\)](mailto:@luc.edu) email and send to shelquist@luc.edu with only "CHEM 180" in the subject line of the message, OR, reply directly to an email that I've sent to the class via Sakai. Doing either of these will ensure that I read your message. In most cases I will be able to respond within 24 hours Monday-Friday when classes are in session. You are encouraged to use Office Hours to get immediate answers to your questions, and to use your classmates as resources for help. You are welcome to email me in the evenings/nighttime, and you can expect a response sometime during the next day.

**Office Hours Policy:** *You are welcome to stop by at any time* to see if my door is open and check my posted schedule. Occasional extra hours may be announced in class. For [regular OH](#), just show up!! Bring your questions anytime during the times listed. Bring a classmate with you or meet your classmates there to work together & get feedback & help.

**Office Hours Schedule:** In the [STEM Center](#) St. Joseph Hall, Cafeteria: Mondays 2:30-4:00pm

- In the Flanner 200 office suite: Thursdays 3:00-4:30pm; Fridays 9:00-10:00am
- A limited number of short, individual appointments will be available weekly, by advance sign-up, via Sakai Sign-up page
- Occasional Sunday afternoon hours will be held in Ireland's, see [Sakai Resources for Help](#) for weekly updates

### SI information

There are Supplemental Instruction (SI) study sessions available for this course. SI sessions are led by an SI leader, Katherine Summers, who is a student that has recently excelled in the course. Session attendance is open to all, and while it is voluntary, it is extremely beneficial for those who attend weekly. Times and locations for the SI session can be found here: [www.luc.edu/tutoring](http://www.luc.edu/tutoring). Students who attend these interactive sessions find themselves working with peers as they compare notes, demonstrate and discuss pertinent problems and concepts, and share study and test-taking strategies. Research shows students whom regularly attend sessions have higher grades at the end-of-the-semester and more deeply understand course concepts than those who do not. Students are asked to arrive with their Loyola ID number, lecture notes, and textbook.

### Required Course Materials

- WileyPlus, see [Sakai](#) for additional information and recommendations
- Molecular Model Kit (Duluth Labs MM-005 or equivalent)
- Scientific Calculator (non-programmable, non-graphing, and independent of another device such as a phone or tablet)
- Loyola Sakai course management site: [sakai.luc.edu/portal/](http://sakai.luc.edu/portal/) and tools integrated into the site.
- Loyola email: messages are sent to the entire class via Sakai, linked to your Loyola email account

- Additional web-based systems will be used for uploading your work and facilitating feedback and evaluation. Registration will be free but required. These may include [Gradescope](#) and other sites.
- Additional software may be used. These may include applications that convert photos to PDFs, and collaboration materials for group work.

**Copyright/Intellectual Property reminder:** Course materials provided by your instructors at Loyola, including my materials, may not be shared outside any course without the instructor's **written permission**. Content posted without permission will be in violation of Copyright/Intellectual Property laws.

### Course Description

This course is the second in a sequence of multiple chemistry courses designed to create foundational knowledge and proficiency in essential chemistry concepts and skills. It includes the following topics: acids and bases, buffers, chemical equilibrium, molecular thermodynamics and kinetics, nucleophilic substitutions, elimination reactions, carbonyl compounds and reactions with applications to biochemical pathways. Historical and current developments in chemistry as well as real-world problems that chemists address are incorporated into the course.

Alongside specific content, these themes will cycle through each of the foundational courses. They include:

- Structure-Activity Relationships
- The culture and practice of science.
- Energy.
- Polymers, proteins, and macromolecules.
- Sustainability.
- Chemical synthesis, purification, characterization, and analysis.

### Learning Outcomes

The emphasis of this course is on understanding, prediction, investigation, explanation and evaluation over memorization. This means that students must foster their problem solving skills, ability to make claims based on evidence, use and understanding of models and their limitations, and skills of effective communication of scientific results. It is not enough to know *what* happens in chemistry, the student must also be able to explain *why* it happens. When successful, a student will be able to:

- Qualitatively and quantitatively describe and explain how molecular structure and stability changes over time in chemical reactions.
- Draw and interpret multiple representations of structures depicting reactivity.
- Predict and draw reaction products.
- Propose and draw logical reaction mechanisms.
- Quantify relationships between variables controlling chemical systems.
- Differentiate among closely related factors, categorize problem types, and select appropriate tools to solve these problems.
- Apply chemical principles to explain natural phenomena.

### Class Attendance & Course Coverage

You will have the chance to introduce yourself to multiple classmates early in the course. Our actual pace may vary from this schedule: if you miss a class for any reason, it is your responsibility to work through the content, and I also suggest you contact a classmate for further discussion of the topics as you are still responsible for all material covered and assigned. An outline will be shown at the beginning of each class, but I do not have published lecture notes. Lectures will be presented using some pre-recorded content to be viewed on Panopto (via Sakai) to supplement "live" lectures & discussions. Slides/handouts/links/animations and other additional resources will be shared on Sakai. We are covering the course topics in a more traditional (structure-first) order compared to how topics are listed in your textbook. To help you stay on track, pre-lecture readings will be posted and continually updated on the Sakai overview page for this course. Required Pre-lecture Homework objectives are assigned and continually updated in WileyPlus. Post-lecture Highly Recommended problems for additional daily practice will be continually updated online. We will not cover every topic in every chapter of the textbook this semester. Focus first on the material that is directly covered in lecture and assigned or recommended. Explore the additional material in the textbook for your own interest and enrichment.

### Classroom & Group Work Guidelines

The classroom is a space designed for learning. My expectations are that all voices will be heard and appreciated in the classroom, and that we will invite each other to engage while recognizing that contributions

can take multiple forms. You will write expectations/guidelines for your group work this semester: this will be an essential part of the course.

### **Student and Faculty Expectations**

I expect you to take ownership of your learning and to use office and SI sessions as learning resources to help you reach your desired level of achievement in the course. For this course, it is anticipated that the average independent working time (outside of class) required to learn the material in order to achieve a minimal passing grade of C- is 1-2 hours per day, every day, but your needs will also vary depending on your prior knowledge and ability to master cumulative concepts in the course material as the semester progresses. What can you expect of me? My primary objectives are to provide you with the tools, environment, encouragement, and support to learn Chemistry. Because the course objectives are based on what students will learn, my teaching techniques include the use of pre-lecture homework, active learning and metacognition, to help you maximize your learning. I expect that all of us will work together!

### **Student Accommodations**

Loyola University Chicago provides reasonable accommodations for students with disabilities. Any student requesting accommodations related to a disability or other condition is required to register with the Student Accessibility Center (SAC). Professors will receive an accommodation notification from SAC, preferably within the first two weeks of class. Students are encouraged to meet with their professor individually in order to discuss their accommodations. All information will remain confidential. Please note that in this class, software may be used to audio record class lectures in order to provide equitable access to students with disabilities. Students approved for this accommodation use recordings for their personal study only and recordings may not be shared with other people or used in any way against the faculty member, other lecturers, or students whose classroom comments are recorded as part of the class activity. Recordings are deleted at the end of the semester. For more information about registering with SAC or questions about accommodations, please contact SAC at 773-508-3700 or [SAC@luc.edu](mailto:SAC@luc.edu).

### **Course Repeat Rule**

Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). The Department advises that it is preferable to complete a course with a grade of C or C-, and to demonstrate growth in future coursework, than to withdraw from a course.

After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of Chemistry & Biochemistry website: <https://www.luc.edu/chemistry/forms/> and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

### **Academic Integrity**

All students in this course are expected to have read and to abide by the demanding standard of personal honesty, drafted by the College of Arts & Sciences, which can be viewed at:

<https://www.luc.edu/cas/advising/academicintegritystatement/>

A basic mission of a university is to search for and to communicate the truth as it is honestly perceived. A genuine learning community cannot exist unless this demanding standard is a fundamental tenet of the intellectual life of the community. Students of Loyola University Chicago are expected to know, to respect, and to practice this standard of personal honesty.

Academic dishonesty can take several forms, including, but not limited to cheating, plagiarism, copying another student's work, submitting false documents, and deliberately disrupting the performance of other class members. Standards apply to both individual and group assignments.

Regarding the use of Artificial Intelligence: our Provost has expressed to "Let us all make sure we are learning and sharing best practices and not allowing AI to do the learning for us." In this course, any work you submit for credit must represent your own ideas and understanding of the assigned material. If you are uncertain about any case where your use of AI may be in conflict with University or course standards, please see me to discuss your concerns.

An instance of academic misconduct (including those detailed on the website provided above or in this syllabus) will be reported to the Department Chair and the academic Dean's office. You will receive a score of zero on the item in question for any instance of academic misconduct, and that score cannot be dropped.

**Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC):**

Students missing classes while representing Loyola University Chicago in an official capacity (e.g., intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes.

Students should discuss with faculty the potential consequences of missing lectures and the ways in which they can be remedied. Students must provide their instructors with proper documentation i.e., "[Athletic Competition & Travel Letter](#)" describing the reason for and date of the absence.

This documentation must be signed by an appropriate faculty or staff member and it must be provided to the professor in the first week of a semester. It is the responsibility of the student to make up any assignments. If the student misses an examination, the instructor is required to allow the student to take the examination at another time.

(<https://www.luc.edu/athleteadvising/attendance.shtml>)

Students who will miss class for an academic competition or conference must provide proper documentation to their instructor as early in the semester as possible.

**Accommodations for Religious Reasons**

If you have observances of religious holidays that will cause you to miss class or otherwise effect your performance in the class you must alert the instructor **within 10 calendar days of the first class meeting of the semester** to request special accommodations, which will be handled on a case by case basis.

**Other Items**

- A link to the official Loyola calendar can be found here: <https://www.luc.edu/academics/schedules/>
- The Withdraw deadline for the semester is on Friday, November 3.
- Loyola is using SmartEvals to provide instructor & course feedback. OIE will send emails near the end of the term.
- Additional resources, advice, and suggestions for success (from multiple sources) will be posted/updated on Sakai.

**Class Recording & Content Information**

In general lecture, meetings may be recorded. The following is a mandatory statement for all courses in the College of Arts & Sciences (CAS). We will discuss class norms and standards during the first week and continue the discussion as needed throughout the semester.

**Privacy Statement**

Assuring privacy among faculty and students engaged in online and face-to-face instructional activities helps promote open and robust conversations and mitigates concerns that comments made within the context of the class will be shared beyond the classroom. As such, recordings of instructional activities occurring in online or face-to-face classes may be used solely for internal class purposes by the faculty member and students registered for the course, and only during the period in which the course is offered.

Students will be informed of such recordings by a statement in the syllabus for the course in which they will be recorded. Instructors who wish to make subsequent use of recordings that include student activity may do so **only** with informed written consent of the students involved or if all student activity is removed from the recording. Recordings including student activity that have been initiated by the instructor may be retained by the instructor only for individual use.

**Additional Content, Copyright & Intellectual Property Statement**

By default, students may not share any course content outside the class without the informed written consent of the owner of that content. This includes any additional recordings posted by students, materials provided by the instructor, and publisher-provided materials. For example, lectures, quiz/exam questions, book figures/slides, and videos may not be shared online outside the class. In some cases, copyright/IP violations may overlap with breaches of academic integrity. Remember that obtaining consent to share materials is an active process.

**Pass/Fail Conversion Deadlines and Audit Policy**

A student may request to convert a course into or out of the "Pass/No-Pass" or "Audit" status only within the first two weeks of the semester. For the Fall 2023 semester, students can convert a class to "Pass/No-Pass"

or "Audit" through Monday, September 11th. Students must submit a request for Pass/No-Pass or Audit to their Academic Advisor.

### Final Exam

The University sets the schedule for all final exams. The final will be held on:

**Thursday, December 14<sup>th</sup>, 7:00 PM**

You will have exactly 2 hours to complete the exam. Additional time will not be granted, even if you start late. There will be no make-up final exams given under any circumstance, and the exam will not be given early, either.

Instructors may not reschedule final exams for a class for another day and/or time during the final exam period. There can be no divergence from the posted schedule of dates for final exams. Individual students who have four (4) final examinations scheduled for the same date may request to have one of those exams rescheduled. If a student reports having four final examinations scheduled for the same date, students should be directed to e-mail a petition to Adam Patricoski, Assistant Dean for Student Academic Affairs, CAS Dean's Office (apatricoski@luc.edu).

### Universal Absence Accommodation Policy

The purpose of a universal absence accommodation policy is to account for emergency circumstances (e.g., serious illness, caring for a family member, car accident) that require you to be absent from class, while maintaining fairness in grading for students who attend and complete all in-class graded assignments. We believe that class attendance and participation are essential for your success in this class, and that your health is important to us and our shared community. Please use good judgement and stay home if necessary/prudent for your circumstances.

This is the universal accommodation policy for in-class graded assignments:

- Group Work: the specification for an A is  $\geq 12$  completions to allow at least 2 missed quizzes.
- FOs: multiple attempts at Mastery are automatically provided during the term.
- COs: you are eligible to submit for Proficiency after the first attempt at an CO whether you complete the problems or not; reattempts at Mastery are available during the term.

You may provide documentation for an absence, but it is not required. These accommodations are automatically available to all students.

### Course Topics

We will not cover every topic in every chapter of the custom etext this semester, but the material will usually come from the Chapters listed below. Focus first on the material that is directly covered in classes and assigned or recommended. Explore the additional material in the textbook for your own interest and enrichment.

Chapter 1: Introduction to Structure and Reactivity

Chapter 2: Functional Groups and Bronsted-Lowry Acids and Bases

Chapter 3: Introduction to Equilibrium

Chapter 4: Acid-Base Equilibria and Buffers

Chapter 5: Thermodynamics

Chapter 6: Chemical Kinetics

Chapter 7: Introduction to Reactions and Mechanisms

Chapter 8: Nucleophilic Substitution Reactions and Elimination Reactions

Chapter 9: Alkanes and Cycloalkanes

Chapter 10: Stereoisomerism and Reactions

Chapter 11: Introduction to Carbonyl Chemistry and Nucleophilic Addition Reactions

Chapter 12: Carbohydrates

Chapter 13: Carboxylic Acids and Their Derivatives: An Introduction to Nucleophilic Acyl Substitution Reactions

Chapter 14: Lipids, Amino Acids and Peptides

***• Grading information is on the next pages, and I hope that the measure of what you gain from this course will include much more than the letter on your transcript. Your success in this course is important to me. Let me know what I can do to help you meet your needs and fulfill the requirements to succeed. – Dr. Helquist***

## Course Grading System Design

There are three basic principles that we have used to design the grading system for this course. These are for you to:

1. Understand what the standards and requirements are for each letter grade so that you can choose what level of academic achievement to pursue in this course. We encourage each of you to strive for high achievement because we believe in the potential of all students to learn and improve their abilities in chemistry.
2. Expect a challenging but flexible learning environment. The standards for demonstrating your Mastery of the course material are high in each area, but the methods for meeting the standards are designed to give you multiple chances to revise and improve the quality of your work throughout the semester.
3. Learn from mistakes. Deep, connected learning involves hard work and reflection on your progress. Chemistry is a cumulative subject where the new topics build on prior knowledge and this system is designed for cycles of learning.

### Standards

The standards for each letter grade are listed here according to all required course components, listed in columns. You must meet or exceed all of the standards listed to earn the corresponding letter grade: standards are not averaged across components. These lists are intended for complete transparency: you do not need to do any extra work to figure out what is required for any grade, and we will revisit the standards and expectations after the early rounds of testing to help you gauge your progress in the course. Grades are only based on the criteria listed in the syllabus: no substitutions, and no additions. Descriptions of the components are found on the following pages.

#### A Standards

WileyPlus:  $\geq 90\%$  Overall Grade  
 Group Quiz:  $\geq 12$   
 FO Mastery:  $\geq 34$   
 CO Mastery:  $\geq 16 + 2$  Proficiency  
 Project: A-Spec

#### B Standards

WileyPlus:  $\geq 80\%$  Overall Grade  
 Group Quiz:  $\geq 11$   
 FO Mastery:  $\geq 28$   
 CO Mastery:  $\geq 11 + 3$  Proficiency  
 Project:  $\geq$  B-Spec

#### C Standards

WileyPlus:  $\geq 70\%$  Overall Grade  
 Group Quiz:  $\geq 9$   
 FO Mastery:  $\geq 22$   
 CO Mastery:  $\geq 5 + 5$  Proficiency  
 Project:  $\geq$  C-Spec

#### A- Standards

WileyPlus:  $\geq 90\%$  Overall Grade  
 Group Quiz:  $\geq 12$   
 FO Mastery:  $\geq 32$   
 CO Mastery:  $\geq 14 + 3$  Proficiency  
 Project: A-Spec

#### B- Standards

WileyPlus:  $\geq 80\%$  Overall Grade  
 Group Quiz:  $\geq 80\%$   
 FO Mastery:  $\geq 26$   
 CO Mastery:  $\geq 9 + 5$  Proficiency  
 Project:  $\geq$  B-Spec

#### C- Standards

WileyPlus:  $\geq 70\%$  Overall Grade  
 Group Quiz:  $\geq 9$   
 FO Mastery:  $\geq 20$   
 CO Mastery:  $\geq 4 + 4$  Proficiency  
 Project:  $\geq$  C-Spec

#### B+ Standards

WileyPlus:  $\geq 80\%$  Overall Grade  
 Group Quiz:  $\geq 11$   
 FO Mastery:  $\geq 30$   
 CO Mastery:  $\geq 13 + 3$  Proficiency  
 Project:  $\geq$  B-Spec

#### C+ Standards

WileyPlus:  $\geq 70\%$  Overall Grade  
 Group Quiz:  $\geq 10$   
 FO Mastery:  $\geq 24$   
 CO Mastery:  $\geq 7 + 5$  Proficiency  
 Project:  $\geq$  C-Spec

#### D Standards

WileyPlus:  $\geq 50\%$  Overall Grade  
 Group Quiz:  $\geq 6$   
 FO Mastery:  $\geq 12$   
 CO Proficiency:  $\geq 6$

Note: a student who fails to meet the standards for a grade of D will receive a grade of F for the course.

### Posting of Grades

Final course grades at the end of the semester are posted only LOCUS. Grades are never sent via email. WileyPlus scores are automatically recorded in the WileyPlus Gradebook for that system. Scores for all other required components will be made available on Sakai. Each student will see an estimated midterm grade in LOCUS before the withdraw deadline.

### Course Assessment

All of the following are required components of your course grade:

**WileyPlus: Required Homework**

Registration information is on Sakai; use of this system includes eText access. The purpose of these assignments is to help you keep up with the course material by preparing ahead for each class. You will get as much benefit from these assignments as you choose to put forth in your effort to solve the problems on your own: a list of reading (textbook sections) will be continually updated to correlate with the WileyPlus pre-lecture assignments. Typically, you will have 2-3 required assignments per week, always due at 11:59pm, posted at least 24 hours in advance. Assignments will be submitted completely online with the individual grading policy listed with each assignment. Contact [Wiley Support](#) for help with technical aspects of using WileyPlus. Additional practice assignments will be posted that will not count toward the point total for your course grade.

**Group Quizzes**

On average, 1 quiz per week, usually completed in assigned groups. Most assignments will be completed in class and submitted to Gradescope. The purpose of participation is to improve your learning by: 1) cooperation, communication and support among your classmates as you practice the skills required for success in the course; and 2) providing feedback on your progress to encourage reflection and improvement. Quizzes will include test questions from previous semesters. You will get as much benefit from these quizzes as you choose to put forth in your effort and you are expected to correct your work after receiving feedback. Each quiz will contribute equally toward this category in your course grade. Refer to the Universal Absence Accommodation Policy for missed quizzes.

**Foundational Objectives (FOs): Mastery Testing**

The purpose of testing is to align your course grade with your level of learning, based on your mastery of Foundational Objectives (FOs). The FOs are all related to the Course Content & Learning Outcomes on the first page of this syllabus. A list of FOs will be updated for each unit as we progress through the material. Questions will be scored as Mastered or Not Mastered for each FO. A score of Mastered is earned for correctness and completeness of the problem(s), and each FO may only be counted once toward your FO Mastery Standard. You will have multiple chances to demonstrate mastery of all the FOs during the term: for example, if you receive a score of Not Mastered for any FO on the first test (or if you choose not to attempt an FO), you can try again to earn a score of Mastered for that FO on the second test. Revision of work that does not meet mastery standards is expected for your learning. Because you will have more than one chance to master the FOs, you will also be able to choose which FOs to work toward for the course. Note that the standards for earning Mastery will be high: by definition, there is no partial credit, but you will learn the standards from the examples for class activities. Rounds of testing are scheduled for **September 12, October 3, October 26, November 16, December 7**, with an additional round scheduled during the final exam period. Specific FO dates and timing will be announced at least one week in advance. All procedures, allowed resources, and requirements will be posted before each round of testing.

**Comprehensive Objectives: Mastery Testing & Proficiency Revisions**

The purpose of testing is to align your course grade with your level of learning, based on your mastery of comprehensive topics. The purpose of COs is to allow you to demonstrate your higher-level skills of applying and analyzing, requiring you to go beyond memorization of facts and processes and transfer your understanding of essential course concepts to new scenarios. The COs are all related to the Course Content & Learning Outcomes on the first page of this syllabus. A list of COs will be updated for each unit as we progress through the material. COs will be scored as Mastered or Not Mastered. A score of Mastered is earned for correctness and completeness of the problem(s). Note that the standards for earning Mastery will be high: by definition there is no partial credit, but you will learn the standards from the examples for class activities. Each round of testing on COs will be followed by an opportunity to resubmit work to earn a score of Proficient for an CO that was Not Mastered in the first testing opportunity. Resubmissions for Proficiency will also earn reattempts of COs. Reattempts will take place with the next round of testing. Note that your grade will not count both Mastery and Proficiency for the same item; a CO that is scored Proficient and then is subsequently Mastered on a re-attempt will count only as being Mastered. Rounds of testing are scheduled for **September 12, October 3, October 26, November 16, December 7**, with an additional round scheduled during the final exam period. Specific CO dates and timing will be announced at least one week in advance. All procedures, allowed resources, and requirements will be posted before each round of testing.

**Project**

Specifications and dates will be provided, beginning in October.

**Changes to Syllabus**

There may be changes to the syllabus during the semester. ***You are responsible for all syllabus changes made in class whether or not you attend.***